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#### Key indicators

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.005 Å R factor = 0.060 wR factor = 0.176 Data-to-parameter ratio = 17.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 1-*tert*-Butyl-2-(4-ethylbenzoyl)-1-(2-mercapto-3,5-dimethylbenzoyl)hydrazine

In the title compound,  $C_{22}H_{28}N_2O_2S$ , the carbonyl group closer to the thiol group is almost perpendicular to the attached benzene ring. The crystal packing is stabilized by intermolecular N-H···O and C-H···O hydrogen bonds.

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### Comment

The 1-tert-butyl-1,2-diacylhydrazines, a new class of insectgrowth regulators, have been found to mimic the action of 20hydroxyecdysone to activate the ecdysone receptor, leading to lethal premature moulting (Wing, 1988, 1995; Wing et al., 1988). Among nonsteroidal ecdysone agonists, 1-tert-butyl-2-(4-ethylbenzoyl)-1-(3,5-dimethylbenzoyl)hydrazine (tebufenozide, RH-5992) has been the first to be commercialized as a lepidopteran-specific insecticide, with a low toxicity profile towards mammals, birds and fishes, as well as towards nontarget arthropods such as insect pollinators, predators, and parasitoids (Dhadialla & Jansson, 1999). At present, another three new structural analogues, namely methoxyfenozide (RH-2485), halofenozide (RH-0345) and chromafenozide (ANS-118), have been brought on the market (Carlson et al., 2001; Yanagi et al., 2000). Therefore, in a search for new insectgrowth regulators with improved biological properties and a different activity spectrum, we synthesized the title compound, (I) (Fig. 1).



In (I), all bond lengths and angles (Table 1) are normal (Allen *et al.*, 1987). In the molecule, the carbonyl group closer to the thiol group is almost perpendicular to the attached benzene ring; the torsion angle O1-C1-C2-C3 is 73.8 (4)° (Table 1). This unusual conformation is mainly caused by the steric bulk of the *tert*-butyl group attached to atom N1. The crystal packing (Fig. 2) is stabilized by intermolecular N-H···O and C-H···O hydrogen bonds (Table 2).

## **Experimental**

To a stirred solution of sulfur dichloride (0.08 mol) and dichloromethane (15 ml) was added a solution of pyridine (0.008 mol) in

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## organic papers

dichloromethane (5 ml) dropwise at 263 K. A solution of 1-tert-butyl-1-(3,5-dimethylbenzoyl)-2-(4-ethylbenzoyl)hydrazine (0.007 mol) in dichloromethane (5 ml) was then added at 263 K. The mixture was stirred at room temperature for 4 h and then poured onto ice. The solid was then filtered off and the filtrate was concentrated under vacuum. The residue was purified by column chromatography on silica gel using petroleum ether (60-90), dichloromethane and ethyl acetate (20:1:1 by volume) as the eluents ...

> $D_x = 1.181 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation Cell parameters from 1021

reflections  $\theta = 3.0-21.6^{\circ}$  $\mu=0.17~\mathrm{mm}^{-1}$ T = 293 (2) KBlock, colourless  $0.30 \times 0.24 \times 0.18 \text{ mm}$ 

#### Crystal data

$C_{22}H_{28}N_2O_2S$
$M_r = 384.52$
Monoclinic, $P2_1/n$
$a = 11.656 (3) \text{\AA}$
b = 9.960(3) Å
c = 18.796(5) Å
$\beta = 97.776 \ (5)^{\circ}$
$V = 2162.1 (10) \text{ Å}^3$
Z = 4

#### Data collection

4433 independent reflections
2103 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.057$
$\theta_{\rm max} = 26.4^{\circ}$
$h = -14 \rightarrow 14$
$k = -12 \rightarrow 4$
$l = -23 \rightarrow 23$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.083P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.060$	+ 0.0808P]
$wR(F^2) = 0.176$	where $P = (F_o^2 + 2F_c^2)/3$
S = 0.99	$(\Delta/\sigma)_{\rm max} = 0.001$
4433 reflections	$\Delta \rho_{\rm max} = 0.41 \ {\rm e} \ {\rm \AA}^{-3}$
254 parameters	$\Delta \rho_{\rm min} = -0.27 \ {\rm e} \ {\rm \AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

#### Table 1

Selected geometric parameters (Å, °).

N1-C1 N1-N2	1.347 (3) 1.388 (3)	N2-C14	1.365 (3)
C1-N1-N2-C14	-96.5 (3)	N1-N2-C14-O2	5.6 (4)
N2-N1-C1-O1	-177.1(2)	O2-C14-C15-C16	176.2 (3)
O1-C1-C2-C3	73.7 (4)		

Table 2		
Hydrogen-bonding geometry (	(Å, °).	

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
N2-H2···O1 <sup>i</sup>	0.87 (1)	2.053 (13)	2.893 (3)	162 (3)
$C16-H16\cdots O1^{1}$	0.93	2.44	3.346 (3)	166
$C20-H20\cdots O2^{ii}$	0.93	2.51	3.315 (4)	145

Symmetry codes: (i)  $\frac{1}{2} - x$ ,  $\frac{1}{2} + y$ ,  $\frac{1}{2} - z$ ; (ii) -x, 1 - y, 1 - z.

All C-bound H atoms were placed in calculated positions, with C-H = 0.93–0.97 Å, and included in the final cycles of refinement using a riding model, with  $U_{iso}(H) = 1.2U_{eq}(C)$ . Atom H1, attached to S1, and H2, attached to N2, were located in a difference Fourier map. Atom H2 was refined freely, while for H1 the riding model was used with  $H1-S1 = 1.0 \text{ Å and } U_{iso}(H1) = 1.5U_{eq}(S1).$ 

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve







Packing diagram of (I), showing the intermolecular hydrogen bonds as dashed lines.

structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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